



Technisches Büro &amp; Chemisches Labor

**BODEN**

# Conversion factors of selected nutrients / chemical compounds

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**Instruction:** Multiply the quantity of the given element / compound **by the factor**.

The conversion tables were calculated according to molecular weights and published by **TB Unterfrauner GmbH** (2022). No liability is assumed for the correctness of the conversion factors.

Given	Sought	Factor	Given	Sought	Factor
<b>CALCIUM</b>					
Ca	CaO	1.399	CaO	Ca	0.715
Ca	CaCO <sub>3</sub>	2.497	CaCO <sub>3</sub>	Ca	0.400
Ca	CaSO <sub>4</sub>	3.397	CaSO <sub>4</sub>	Ca	0.294
Ca	CaSO <sub>4</sub> * 2 H <sub>2</sub> O	4.296	CaSO <sub>4</sub> * 2 H <sub>2</sub> O	Ca	0.233
Ca	CaCl <sub>2</sub>	2.769	CaCl <sub>2</sub>	Ca	0.361
Ca	Ca(OH) <sub>2</sub>	1.849	Ca(OH) <sub>2</sub>	Ca	0.541
CaO	CaCO <sub>3</sub>	1.785	CaCO <sub>3</sub>	CaO	0.560
CaO	CaSO <sub>4</sub>	2.428	CaSO <sub>4</sub>	CaO	0.412
CaO	CaCl <sub>2</sub>	1.979	CaCl <sub>2</sub>	CaO	0.505
CaO	Ca(OH) <sub>2</sub>	1.321	Ca(OH) <sub>2</sub>	CaO	0.757

<b>MAGNESIUM</b>					
Given	Sought	Factor	Given	Sought	Factor
Mg	MgO	1.658	MgO	Mg	0.603
Mg	MgCO <sub>3</sub>	3.469	MgCO <sub>3</sub>	Mg	0.288
Mg	MgSO <sub>4</sub>	4.952	MgSO <sub>4</sub>	Mg	0.202
Mg	MgCl <sub>2</sub>	3.917	MgCl <sub>2</sub>	Mg	0.255
Mg	Mg(OH) <sub>2</sub>	2.400	Mg(OH) <sub>2</sub>	Mg	0.417
MgO	MgCO <sub>3</sub>	2.092	MgCO <sub>3</sub>	MgO	0.478
MgO	MgSO <sub>4</sub>	2.986	MgSO <sub>4</sub>	MgO	0.335
MgO	MgCl <sub>2</sub>	2.362	MgCl <sub>2</sub>	MgO	0.423
MgO	Mg(OH) <sub>2</sub>	1.447	Mg(OH) <sub>2</sub>	MgO	0.691

<b>POTASSIUM</b>					
Given	Sought	Factor	Given	Sought	Factor
K	K <sub>2</sub> O	1.205	K <sub>2</sub> O	K	0.830
K	KCl	1.907	KCl	K	0.525
K	K <sub>2</sub> SO <sub>4</sub>	2.228	K <sub>2</sub> SO <sub>4</sub>	K	0.449
K <sub>2</sub> O	KCl	1.583	KCl	K <sub>2</sub> O	0.632
K <sub>2</sub> O	K <sub>2</sub> SO <sub>4</sub>	1.850	K <sub>2</sub> SO <sub>4</sub>	K <sub>2</sub> O	0.541

Given	Sought	Factor	Given	Sought	Factor
<b>NITROGEN</b>					
N	CO(NH <sub>2</sub> ) <sub>2</sub>	2.144		CO(NH <sub>2</sub> ) <sub>2</sub>	N 0.467
N	NO <sub>3</sub>	4.427		NO <sub>3</sub>	N 0.226
N	NH <sub>3</sub>	1.216		NH <sub>3</sub>	N 0.822
N	NH <sub>4</sub>	1.288		NH <sub>4</sub>	N 0.777
N	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	4.717		(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	N 0.212
N	NH <sub>4</sub> NO <sub>3</sub>	2.857		NH <sub>4</sub> NO <sub>3</sub>	N 0.350
N	Ca(CN) <sub>2</sub>	3.288		Ca(CN) <sub>2</sub>	N 0.304
N	N <sub>2</sub> O	1.571		N <sub>2</sub> O	N 0.637

<b>SULFUR</b>					
S	SO <sub>2</sub>	1.998		SO <sub>2</sub>	S 0.501
S	SO <sub>3</sub>	2.497		SO <sub>3</sub>	S 0.401
S	SO <sub>4</sub>	2.996		SO <sub>4</sub>	S 0.334
S	K <sub>2</sub> SO <sub>4</sub>	5.435		K <sub>2</sub> SO <sub>4</sub>	S 0.184
S	MgSO <sub>4</sub> * H <sub>2</sub> O	4.316		MgSO <sub>4</sub> * H <sub>2</sub> O	S 0.232
S	MgSO <sub>4</sub> * 7 H <sub>2</sub> O	7.688		MgSO <sub>4</sub> * 7 H <sub>2</sub> O	S 0.130
S	CaSO <sub>4</sub>	4.246		CaSO <sub>4</sub>	S 0.236
S	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	4.122		(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	S 0.243

<b>PHOSPHORUS</b>					
P	PO <sub>4</sub>	3.066		PO <sub>4</sub>	P 0.326
P	P <sub>2</sub> O <sub>5</sub>	2.291		P <sub>2</sub> O <sub>5</sub>	P 0.436
PO <sub>4</sub>	P <sub>2</sub> O <sub>5</sub>	0.747		P <sub>2</sub> O <sub>5</sub>	PO <sub>4</sub> 1.338
P <sub>2</sub> O <sub>5</sub>	Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.185		Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	P <sub>2</sub> O <sub>5</sub> 0.458

<b>BORON</b>					
B	H <sub>3</sub> BO <sub>3</sub>	5.720		H <sub>3</sub> BO <sub>3</sub>	B 0.175
B	Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> * 10 H <sub>2</sub> O	8.820		Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> * 10 H <sub>2</sub> O	B 0.113
B	B <sub>2</sub> O <sub>3</sub>	3.220		B <sub>2</sub> O <sub>3</sub>	B 0.311

<b>COPPER</b>					
Cu	CuSO <sub>4</sub> * 5 H <sub>2</sub> O	3.929		CuSO <sub>4</sub> * 5 H <sub>2</sub> O	Cu 0.255

<b>MANGANESE</b>					
Mn	MnSO <sub>4</sub> * 4 H <sub>2</sub> O	4.060		MnSO <sub>4</sub> * 4 H <sub>2</sub> O	Mn 0.246
Mn	MnCl <sub>2</sub> * 4 H <sub>2</sub> O	3.602		MnCl <sub>2</sub> * 4 H <sub>2</sub> O	Mn 0.278

Note: No liability is assumed for the correctness of the conversion factors

Given	Sought	Factor	Given	Sought	Factor
<b>MOLYBDENUM</b>					
Mo	$(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} * 4 \text{ H}_2\text{O}$	1.840	$(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} * 4 \text{ H}_2\text{O}$	Mo	0.543
Mo	$\text{Na}_2\text{MoO}_4$	2.146	$\text{Na}_2\text{MoO}_4$	Mo	0.466

<b>SODIUM</b>					
Na	$\text{Na}_2\text{O}$	1.348	$\text{Na}_2\text{O}$	Na	0.742
Na	NaCl	2.542	NaCl	Na	0.393
$\text{Na}_2\text{O}$	NaCl	1.886	NaCl	$\text{Na}_2\text{O}$	0.530

<b>ZINC</b>					
Zn	$\text{ZnSO}_4 * 7 \text{ H}_2\text{O}$	4.398	$\text{ZnSO}_4 * 7 \text{ H}_2\text{O}$	Zn	0.227

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Chemical formula	Designation
<b>CALCIUM</b>	
Ca	Elemental calcium
CaO	Calcium oxide (quicklime)
$\text{CaCO}_3$	Calcium carbonate
$\text{CaCl}_2$	Calcium chloride
$\text{CaSO}_4$	Calcium sulphate
$\text{CaSO}_4 * 2 \text{ H}_2\text{O}$	Calcium sulphate (gypsum)
$\text{Ca(OH)}_2$	Calcium hydroxide (slaked lime)
<b>MAGNESIUM</b>	
Mg	Elemental magnesium
MgO	Magnesium oxide
$\text{MgCO}_3$	Magnesium carbonate (magnesite)
$\text{MgSO}_4$	Magnesium sulphate (epsom salt)
$\text{MgCl}_2$	Magnesium chloride
<b>POTASSIUM</b>	
K	Elemental potassium
$\text{K}_2\text{O}$	Potassium oxide
$\text{K}_2\text{SO}_4$	Potassium sulphate
KCl	Potassium chloride

Chemical formula	Designation
<b>NITROGEN</b>	
N	Elemental nitrogen
CO(NH <sub>2</sub> ) <sub>2</sub>	Urea
NO <sub>3</sub>	Nitrate
NH <sub>3</sub>	Ammonia
NH <sub>4</sub>	Ammonium
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	Ammonium sulphate
NH <sub>4</sub> NO <sub>3</sub>	Ammonium nitrate (KAS, NAC)
Ca(CN) <sub>2</sub>	Calcium cyanide
N <sub>2</sub> O	Nitrous oxide
<b>SULFUR</b>	
S	Elemental sulphur
SO <sub>2</sub>	Sulphur dioxide
SO <sub>3</sub>	Sulphur trioxide
SO <sub>4</sub>	Sulphate
K <sub>2</sub> SO <sub>4</sub>	Potassium sulphate
MgSO <sub>4</sub> * H <sub>2</sub> O	Magnesium sulphate hydrate
MgSO <sub>4</sub> * 7 H <sub>2</sub> O	Magnesium sulphate heptahydrate
CaSO <sub>4</sub>	Calcium sulphate
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	Ammonium sulphate
<b>PHOSPHORUS</b>	
P	Elemental phosphorus
PO <sub>4</sub>	Phosphate
P <sub>2</sub> O <sub>5</sub>	Phosphorus pentoxide
Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	Calcium phosphate
<b>BORON</b>	
B	Elemental boron
H <sub>3</sub> BO <sub>3</sub>	Boric acid
Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> * 10 H <sub>2</sub> O	Borax (sodium tetraborate decahydrate)
B <sub>2</sub> O <sub>3</sub>	Boric anhydride

Chemical formula	Designation
<b>COPPER</b>	
Cu	Elemental copper
CuSO <sub>4</sub> * 5 H <sub>2</sub> O	Copper sulphate pentahydrate
<b>MANGANESE</b>	
Mn	Elemental manganese
MnSO <sub>4</sub> * 4 H <sub>2</sub> O	Manganese sulphate tetrahydrate
MnCl <sub>2</sub> * 4 H <sub>2</sub> O	Manganese chloride tetrahydrate
<b>MOLYBDENUM</b>	
Mo	Elemental molybdenum
(NH <sub>4</sub> ) <sub>6</sub> Mo <sub>7</sub> O <sub>24</sub> * 4 H <sub>2</sub> O	Ammonium heptamolybdate tetrahydrate
Na <sub>2</sub> MoO <sub>4</sub>	Sodium molybdate
<b>SODIUM</b>	
Na	Elemental sodium
Na <sub>2</sub> O	Sodium oxide
NaCl	Sodium chloride (salt)
<b>ZINC</b>	
Zn	Elemental zinc
ZnSO <sub>4</sub> * 7 H <sub>2</sub> O	Zinc sulphate heptahydrate